

Investigation of Autoignition and Combustion Stability of High Pressure Supercritical Carbon Dioxide Oxy-combustion

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Performance period: Oct. 2015 – Sept. 2018

UTSR Project: DE-FE0025174

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Backstory



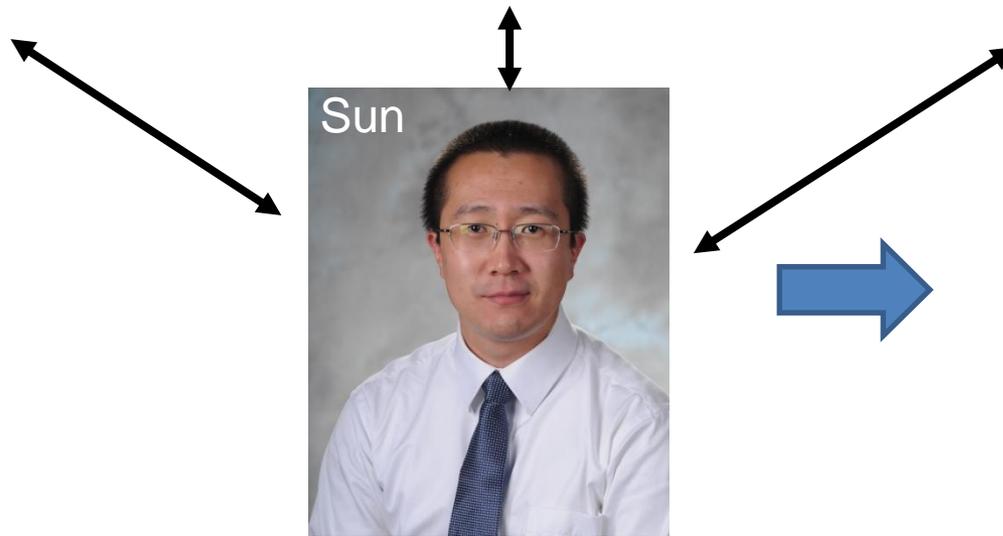
Combustion Dynamics



Shock-tube, SCO_2 System



LES/DNS

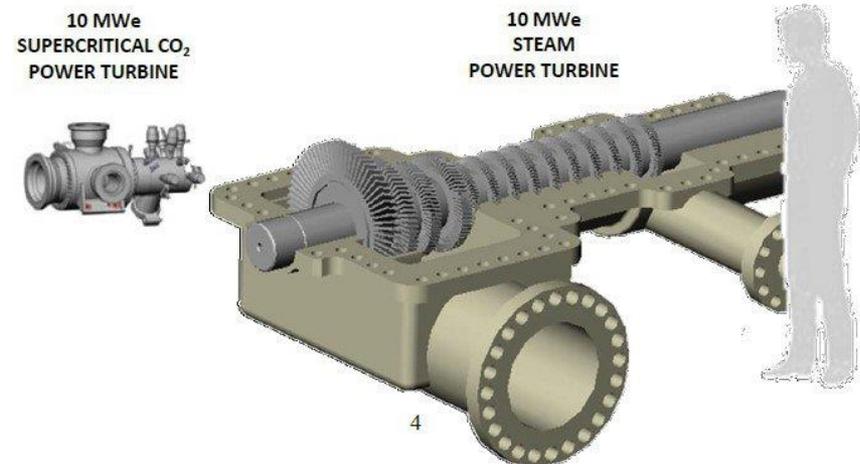


Combustion Chemical Kinetics

Background of Directly Fired Supercritical CO₂ cycle



- High plant conversion efficiencies exceeding 52% (LHV) with ~100% carbon capture
- Lower electricity cost (by ~15%)
- SCO₂ is a single-phase working fluid, and does not create the associated thermal fatigue or corrosion associated with two-phase flow (e.g., steam)
- SCO₂ undergoes drastic density change over small ranges of temperature and pressure → large amount of energy can be extracted → small equipment size



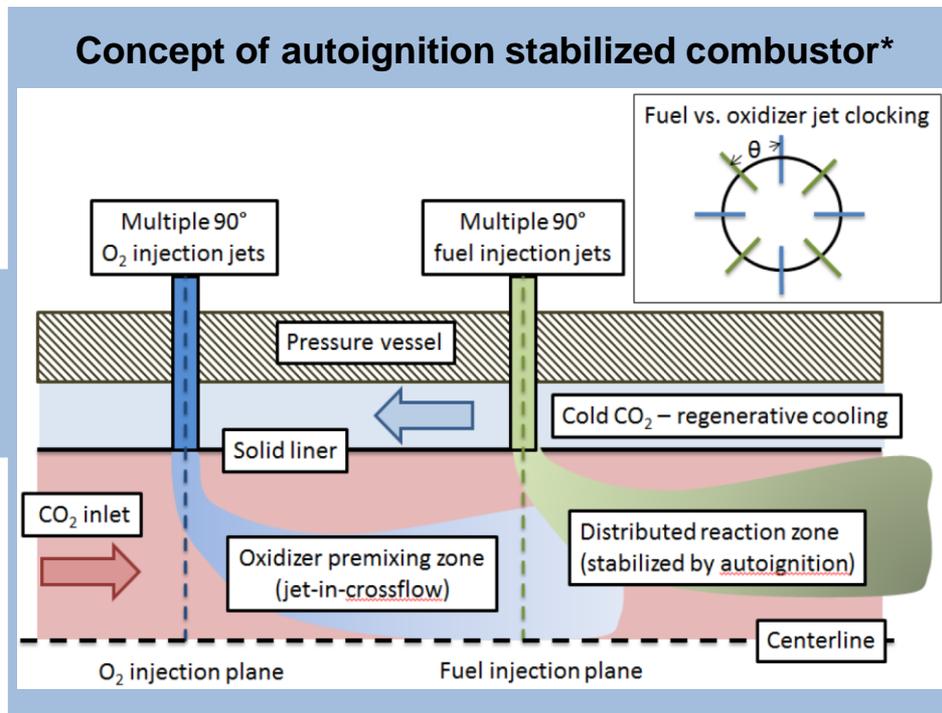
Echogen's 10 MWe sCO₂ power turbine compared to a 10 MWe steam turbine.



Overview of the Scientific Problem

- What fundamental combustion properties/knowledge we need in order to design combustor for SCO_2 oxy-combustion?
- High temperature (~ 1100 K) and high pressure (~ 200 - 300 atm) inlet condition
 - Conventional gas turbine combustor won't work owing to the failure of injector/flame holder at severe thermal environment

Autoignition delays
and
combustion dynamics
of jet in crossflow

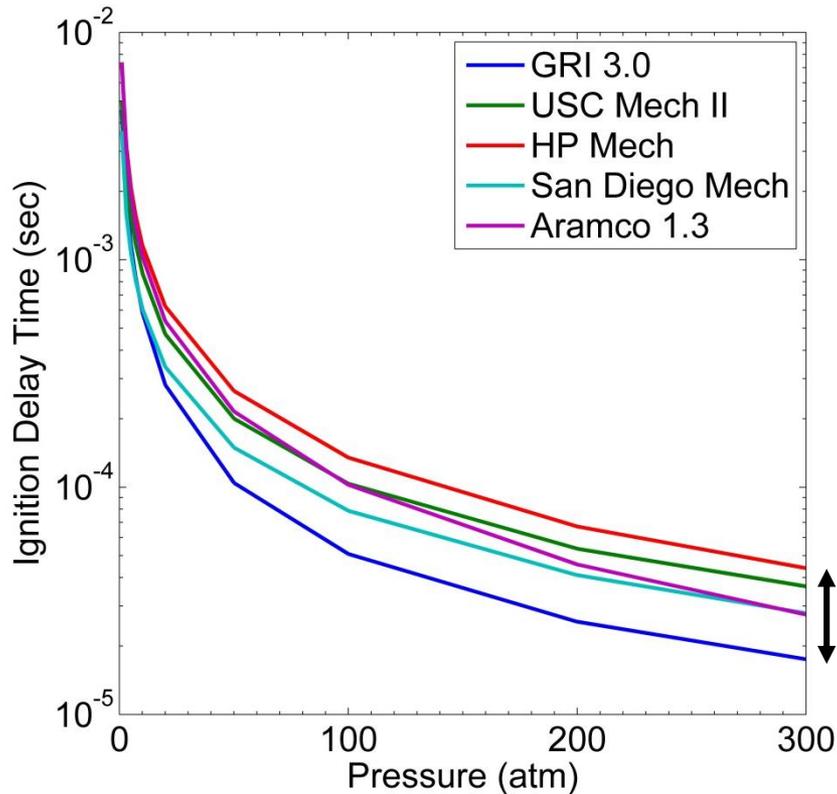


Motivation

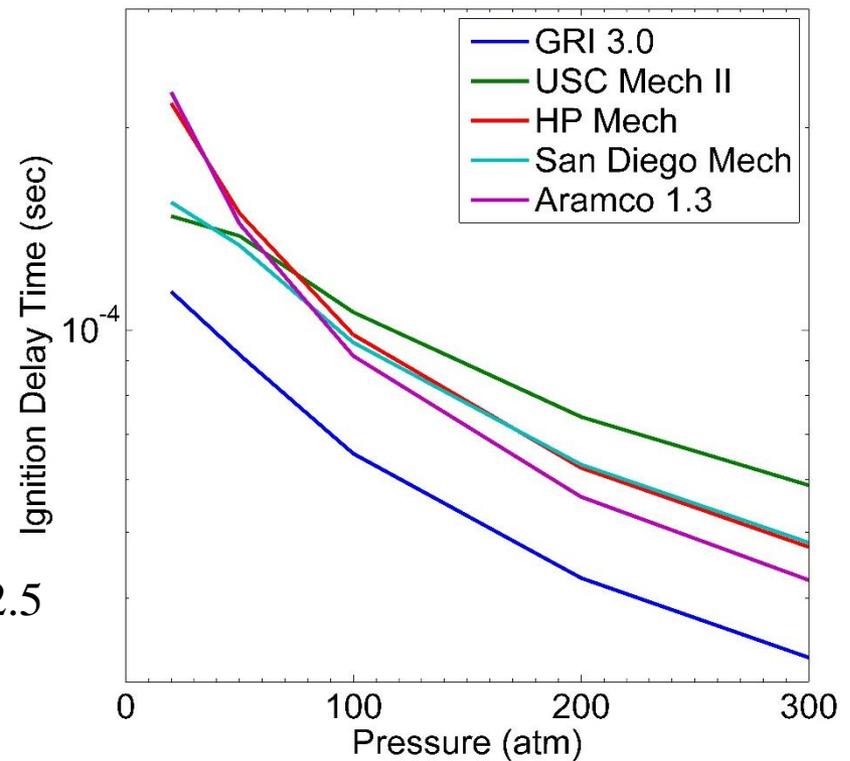


Deviation increases with pressure: knowledge gap
Kinetic models must be validated at regime of interest !!

Predicted autoignition delays from different kinetic models



$\text{CH}_4/\text{O}_2/\text{CO}_2$ (9.5%:19%:71.48%) at 1400 K
from 1 atm to 300 atm



$\text{H}_2/\text{CO}/\text{O}_2/\text{CO}_2$ (14.8%:14.8%:14.8%:55.6%)
at 1200 K from 10 atm to 300 atm

Overview of the Scientific Questions and Proposed Work

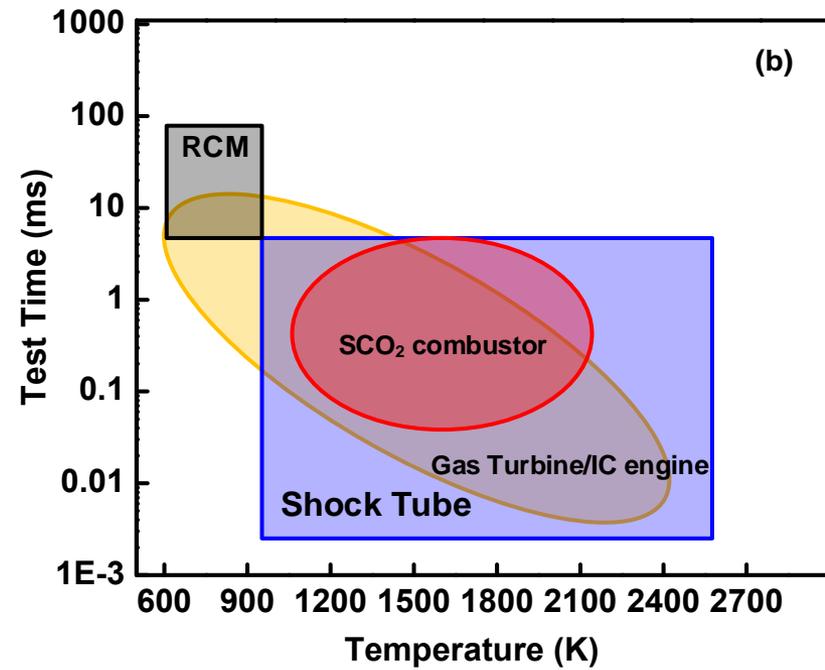
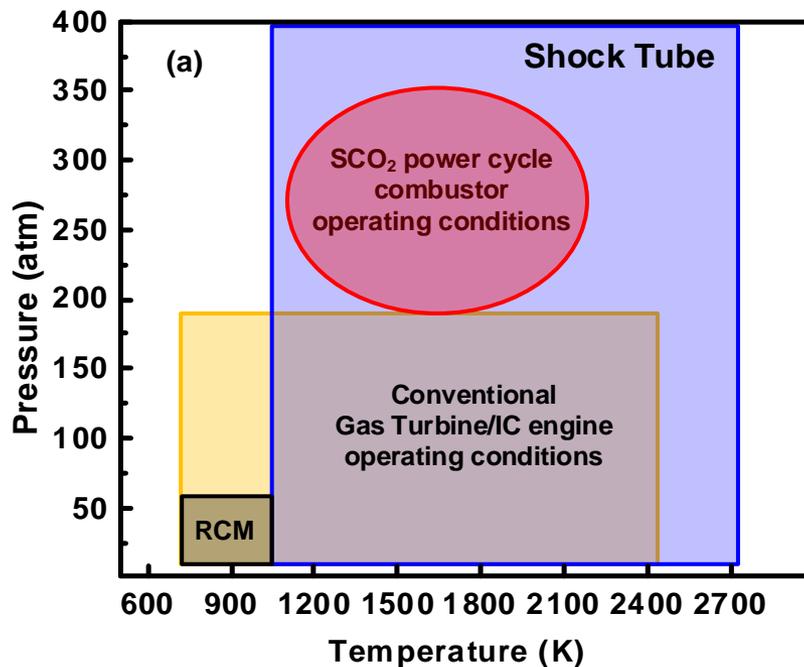


- What is the fundamental combustion properties?
 - Experimental investigation of chemical kinetic mechanisms for SCO_2 Oxy-combustion (Task 1&2: Ranjan & Sun)
- How can we use the mechanism to design combustors?
 - Development of a compact and optimized chemical kinetic mechanism for SCO_2 Oxy-combustion (Task 3: Sun)
- What is the combustor dynamics at this new condition?
 - theoretical and numerical investigation of combustion instability for SCO_2 Oxy-combustion (Task 4&5: Lieuwen, Menon & Sun)

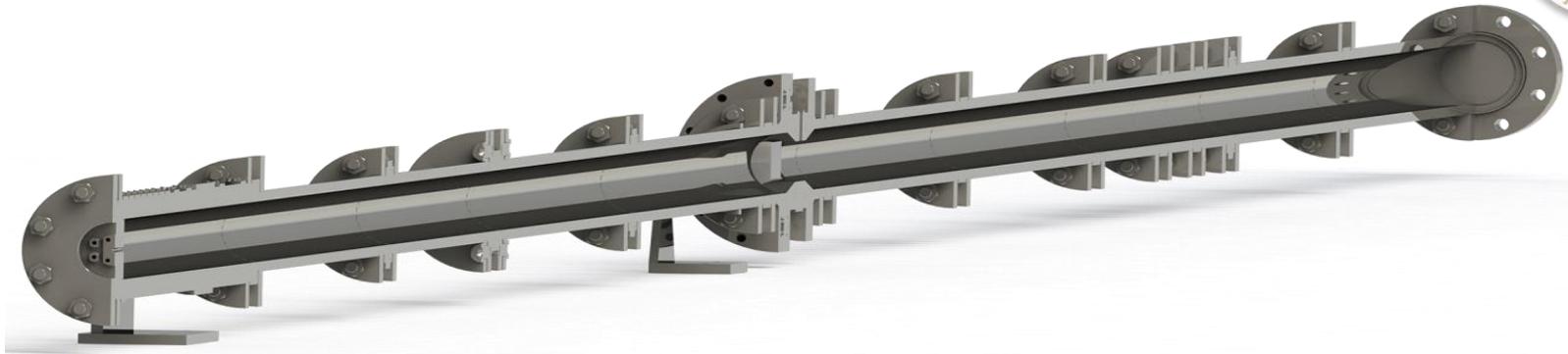
Task 1: Development of a High Pressure Shock Tube for Combustion Studies



- How to study autoignition delays at SCO_2 Oxy-combustion condition?
 - Why Shock-Tube?



Task 1: Development of a High Pressure Shock Tube for Combustion Studies



- Georgia Tech shock tube for fundamental autoignition study is under construction
- Wide pressure range (P up to 300 atm)
- Large ID (152.4 mm) to minimize non-ideal effect at very high pressure condition



Task 1: Development of a High Pressure Shock Tube for Combustion Studies

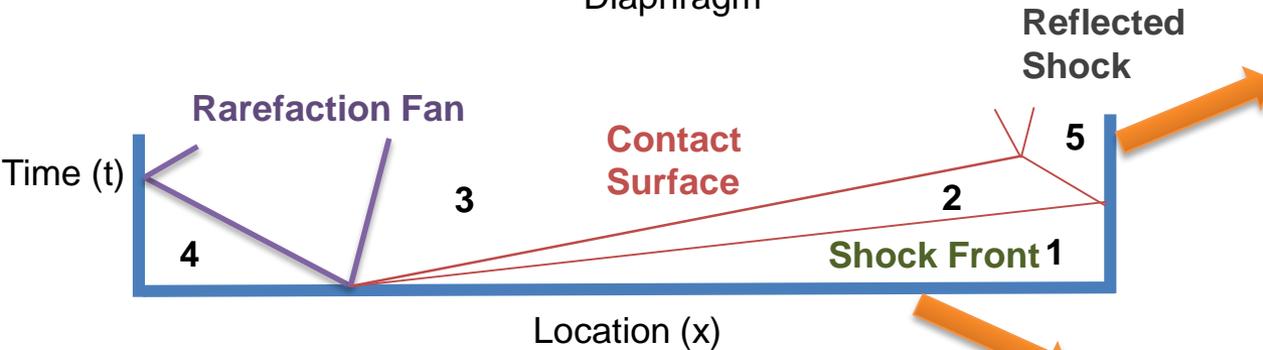


Basics regarding the shock-tube:

Shock Tube Schematic



Diaphragm



Reflected Shock

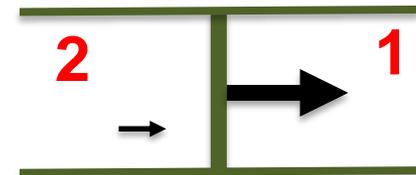
Lab-Frame Reflected Shock



$$T_5 = 1000 - 4000 \text{ K}$$

$$P_5 > P_2$$

Lab-Frame Incident Shock



$$T_2 = 500 - 2000 \text{ K}$$

$$P_2 > P_1$$

Diagnostics: pressure and chemiluminescence

Remind: currently no absorption spectroscopy can work at this condition (above 50 atm)

Task 1: Development of a High Pressure Shock Tube for Combustion Studies



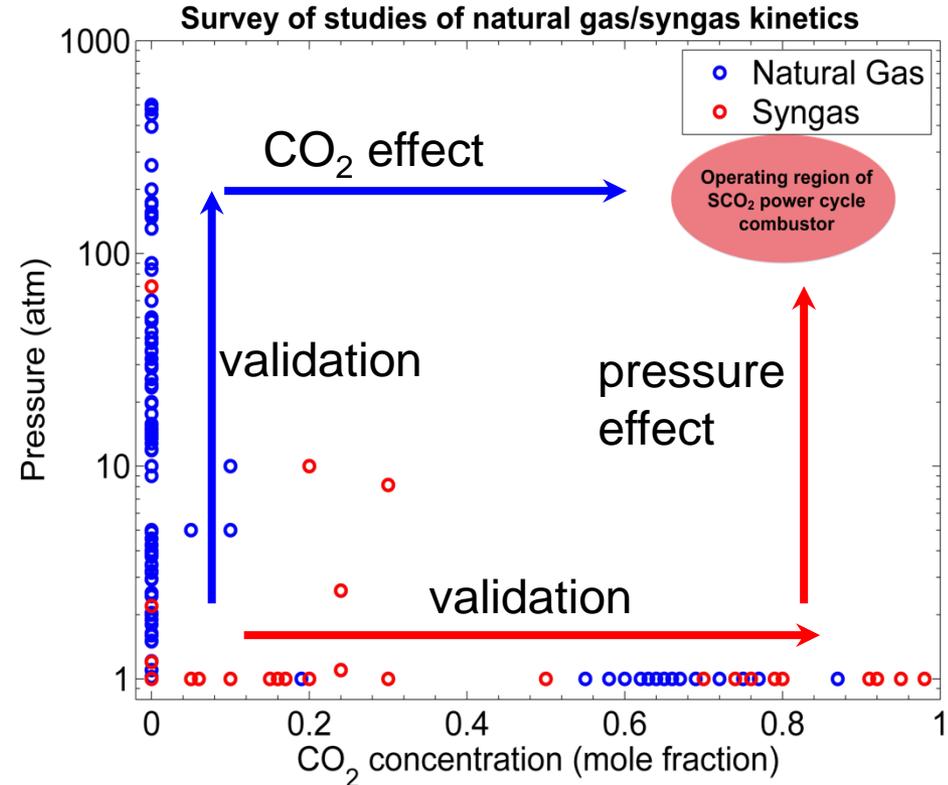
Key Capability of the GT Shock-tube

- Large internal bore (15.24 cm)—to minimize the boundary layer effect (very critical at high pressure conditions)
- It will be long (20 m total)
- Test time 50 ms (can achieve high value with modification of driver gas mixture)
- Diaphragm section replicate the current design in the operational shock-tube for turbulent mixing study
- Test pressure ~300 bar
- Preheating capability
- 0.2 μm or better surface finish
- Optical access from end wall and side-wall
- Several locations for pressure transducers at the end wall and on side wall
- Diagnostic capability to understand the non-ideal effects in the shock-tube

Task 2: Investigation of Natural Gas and Syngas Autoignition in SCO_2 Environment



- Autoignition properties have never been investigated before in region of interest
- This task will investigate critical autoignition properties of natural gas and syngas diluted by CO_2 in region of interest
- Approach for high quality data:
 - Repeat existing experiments for validation
 - Ramp up pressure to study pressure effect
 - Ramp up CO_2 dilute concentration to study CO_2 dilution effect



A new regime to explore!

e.g.:

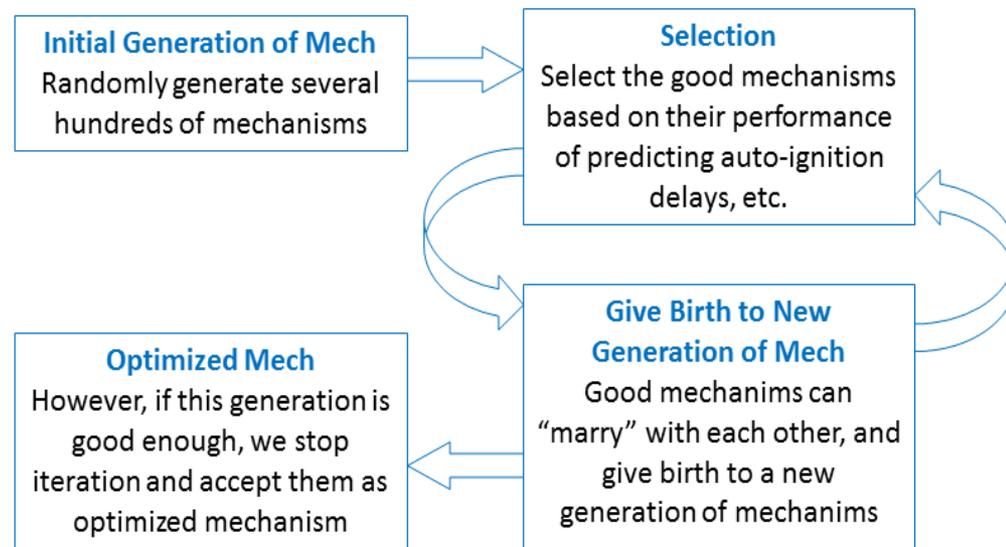
E.L. Petersen, et al, Symp. Combust., 1996(26), 799-806

S. Vasu, et al, Energy Fuels, 2011(25), 990-997

Task 3: Development of a Compact and Optimized Chemical Kinetic Mechanism for SCO_2 Oxy-combustion



- Develop an optimized, validated and compact chemical kinetic mechanism
- Employ the optimized mechanism in LES to study combustion stability
- Approach: optimize chemical kinetic mechanism based on experimental data obtained in task 2.
- Explore other methodology: Bayesian optimization for better optimization



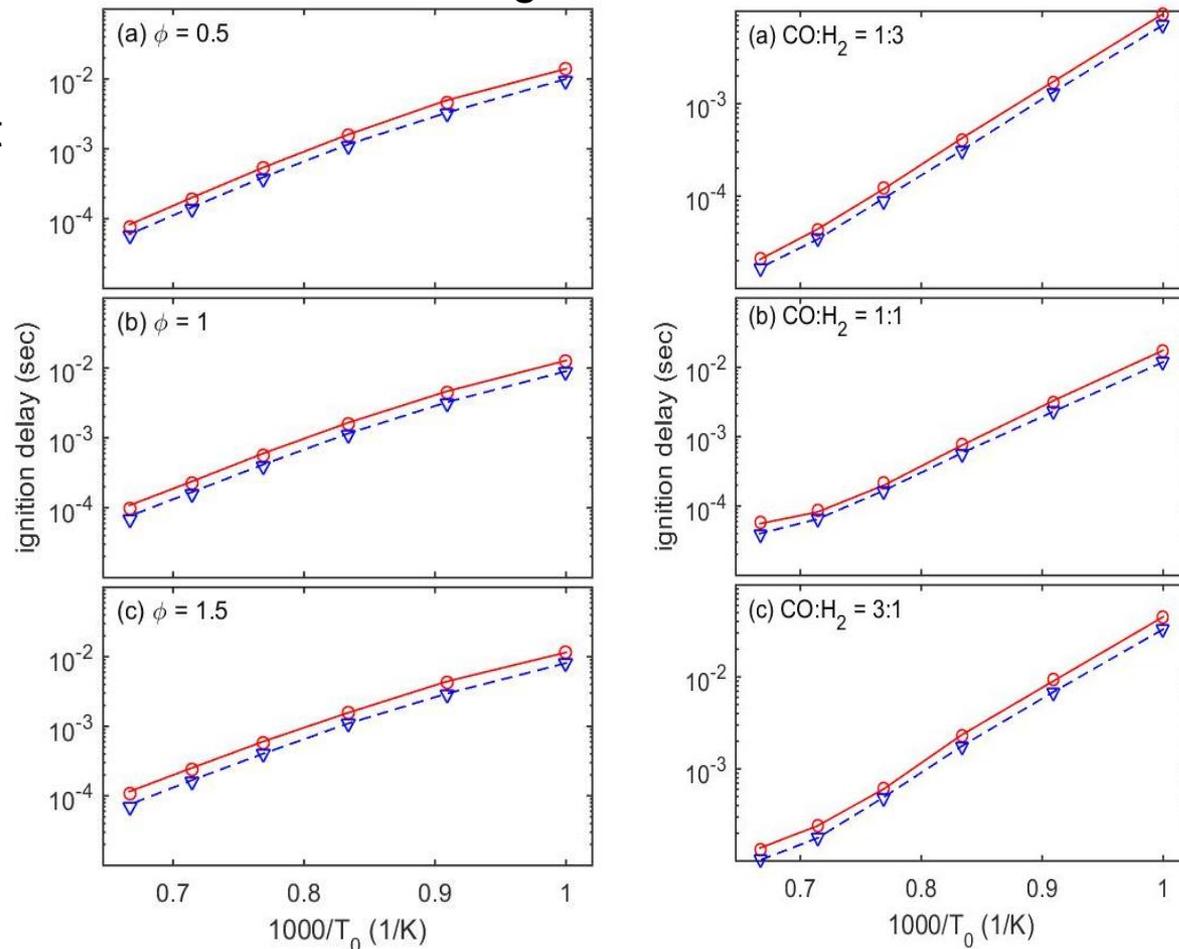
Flow chart of using Genetic Algorithm to optimize chemical kinetic mechanisms

Task 3: Development of a Compact and Optimized Chemical Kinetic Mechanism for SCO_2 Oxy-combustion



- Comparing to existing high pressure autoignition delay data, USC Mech II (111 species) has the best agreement¹. So it is used as a starting point for future optimized mechanism
- A 27 species reduced mechanism² for natural gas ($\text{CH}_4/\text{C}_2\text{H}_6$) and syngas (CO/H_2) is developed
- Comparison of the results from reduced (marker) and detailed mech (line). Solid lines ($p = 200\text{atm}$), dashed line ($p = 300\text{atm}$)

Autoignition



92.5% CO_2 diluted natural gas/ O_2 ($\text{CH}_4:\text{C}_2\text{H}_6=95:5$)

92.5% CO_2 diluted syngas gas/ O_2 ($\phi=1$)

Warning: therm/trans data !!
e.g., CO_2 , different trend

1. A. McClung, DE-FE0024041 Q1FY15 Research Performance Progress Report, SwRI

2. S. Coogan, X. Gao, W. Sun, Evaluation of Kinetic Mechanisms for Direct Fired Supercritical Oxy-Combustion of Natural Gas, TurboExpo 2016

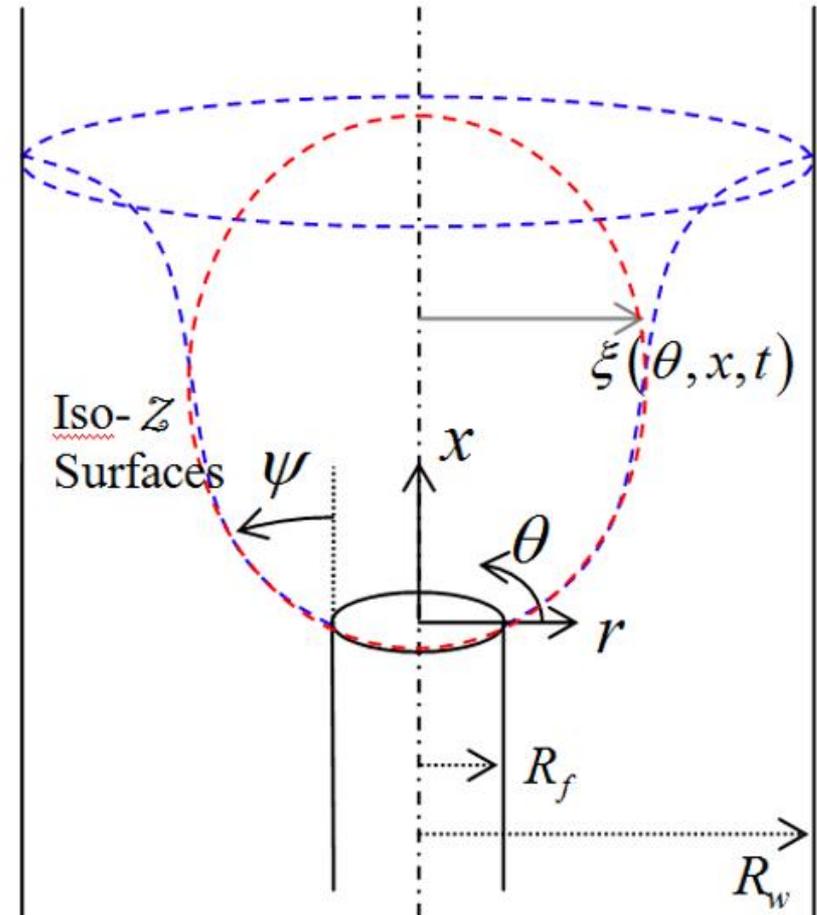
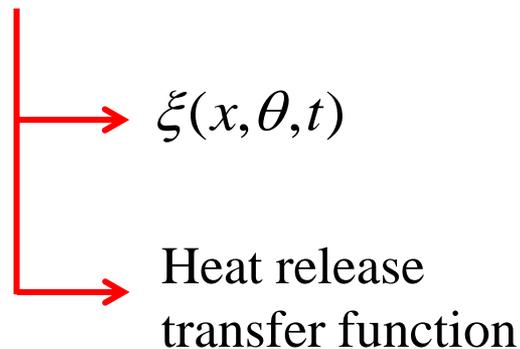
Task 4: Analytical modeling of Supercritical Reacting Jets in Crossflow



- Established model: Mixture fraction formulation

$$\frac{\partial Z}{\partial t} + \vec{u} \cdot \nabla Z = \nabla \cdot (\mathcal{D} \nabla Z)$$

$$Z(x, \theta, \xi(x, \theta, t), t) = Z_{st}$$



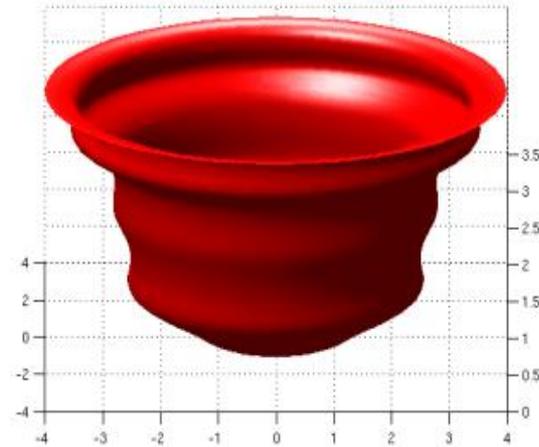
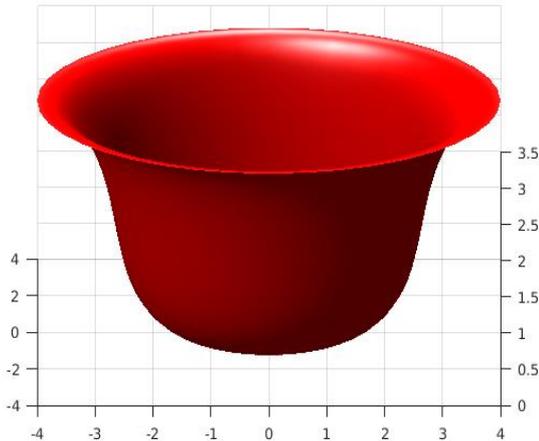
Task 4: Analytical modeling of Supercritical Reacting Jets in Crossflow



• Solution: Space-Time Dynamics of Z_{st} Surface

Bulk Axial Forcing $u_{x,1} = \varepsilon U_0 \exp[-i\omega t]$

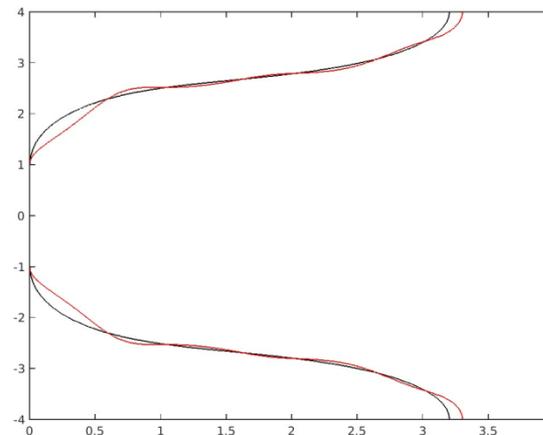
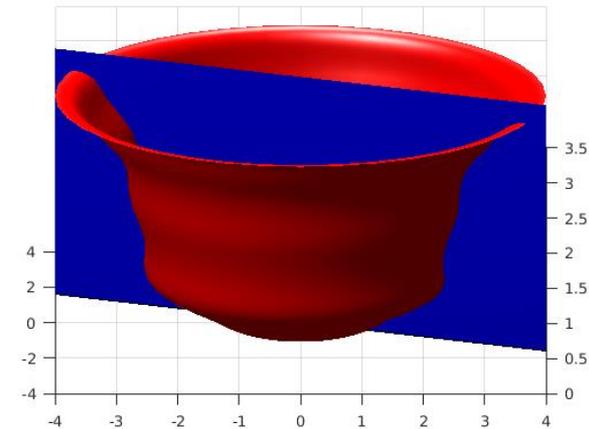
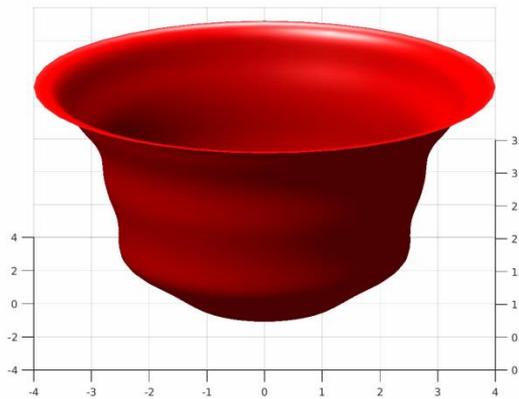
$$Pe \gg 1 \quad \frac{\xi_{1,n}(x,t)}{R_f} = \frac{i\varepsilon \exp[-i\omega t]}{2\pi St} \sin \psi_0(x) \left[1 - \exp\left[2\pi i St \frac{x}{R_f}\right] \exp\left[-\frac{4\pi^2 St^2}{Pe} \frac{x}{R_f}\right] \right] + O\left(\frac{1}{Pe^2}\right)$$



Task 4: Analytical modeling of Supercritical Reacting Jets in Crossflow



- Solution: Space-Time Dynamics of Z_{st} Surface





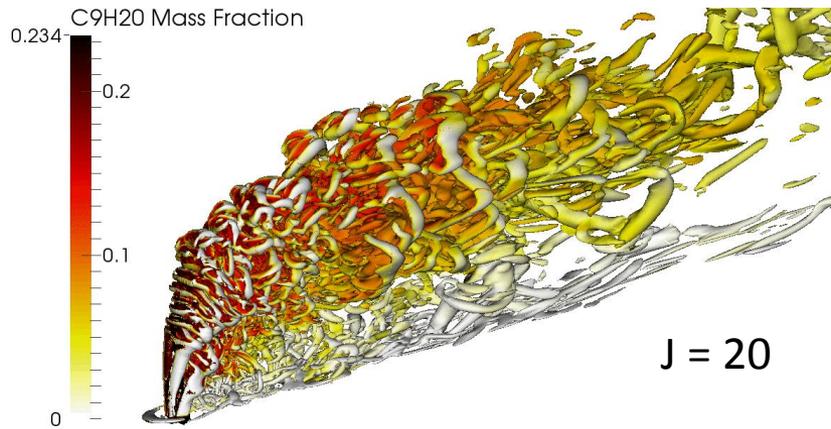
Key Goals of Task 4

- Determine the gain-phase relationship between flow disturbances and heat release oscillations
- Compute time averaged flow and flame features
- Account for supercritical effects on diffusion coefficients, and radiation

Task 5: LES Studies of Supercritical Mixing and Combustion

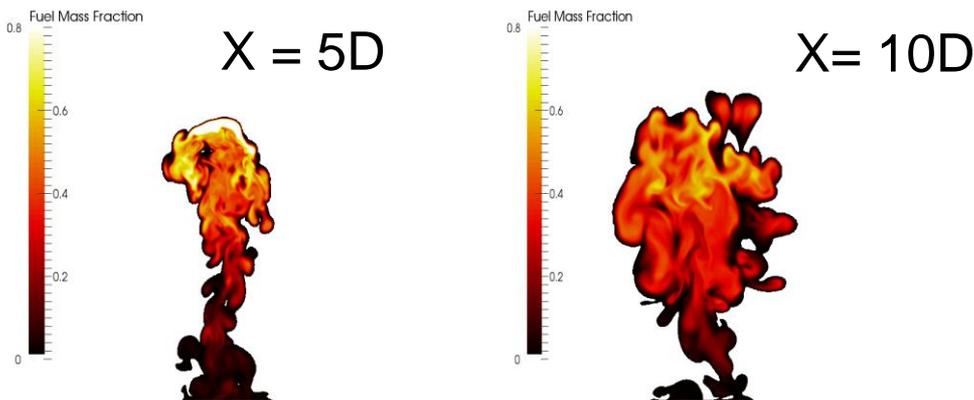


Supercritical Mixing in JICF (leveraged by our rocket engine work)



- LES capability exists to simulate supercritical mixing and reacting flows
- Uses Peng-Robinson EOS for real gas properties with finite-rate kinetics
- Simulations to be used to study mixing and combustion between SCO_2 , fuel/oxidizer
- Effect of radiation

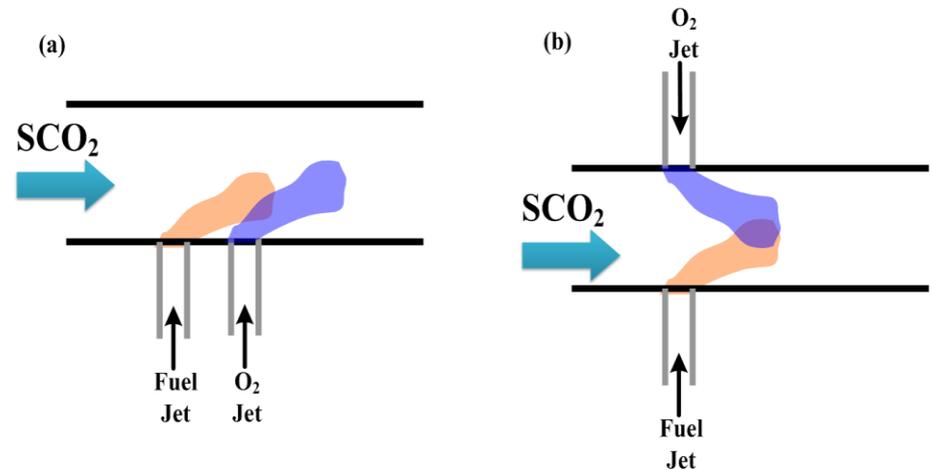
Vorticity Contours of supercritical Kerosene in air



Task 5: LES Studies of Supercritical Mixing and Combustion



- Task 5a: Simulate supercritical mixing/combustion in JICF
- Task 5b: Implement optimized kinetics from Task 3 for reacting studies
- Task 5c: Simulate and analyze conditions resulting in combustion stability in possible combustor geometries
 - Vary inflow and combustor operating conditions
 - Vary injection conditions
- Task 5d: Feedback sensitive reactions to Task 3 to further refine the mechanism



Possible circular combustor design for SCO₂ power cycle (will be modeled)

Task 5: LES Studies of Supercritical Mixing and Combustion



Regime of interest: $P = 200\text{-}300$ atm

fluid	Critical temperature (K)	Critical pressure (atm)
CO ₂	304	72.9
H₂O	647	217.8
CH ₄	190	45.4
C ₂ H ₆	305	48.1
H ₂	32.9	12.8
CO	125.9	34.5
O ₂	154.6	49.7

Warning: Mixing rule !!

A mixture may have one, more than one, or no critical points

$$P_{c, \text{CO}_2} = 72.9 \text{ atm}$$

$$P_{c, \text{C}_{16}\text{H}_{34}} = 25 \text{ atm}$$

$$P_{c, \text{mixture}} = 238 \text{ atm}$$

$$(\text{CO}_2:\text{C}_{16}\text{H}_{34}=0.94:0.06)$$

Transcritical regime exists and is very challenging to model

New physics and chemistry in gas turbine !!



Deliverables

- New fundamental combustion data base for SCO_2 power cycles
- Optimized predictive kinetic mechanism for natural gas and syngas
- Analytic and numerical models of jet in cross flow at SCO_2 power cycle operating conditions

Thank you!
&
Questions?

Acknowledgement:

UTSR Project: DE-FE0025174; PM: Seth Lawson